CHEMISTRY STUDY MATERIALS FOR CLASS 10

(NCERT Based: Questions with Answers)

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CARBON AND ITS COMPOUNDS

Very Short Answer Type Questions (1 Mark)

- 1. Give the names of the following functional groups:
 - (i) —OH (ii) —COOH

Ansswer. (i) Alcohol group (ii) Carboxylic acid group

2. What is the difference in the molecular formula of any two consecutive members of a homologous series of organic compounds?

Answer. —CH₂— is the difference in the molecular formula of any two consecutive members of a homologous series of organic compounds.

3. Name the carbon compound which on heating with excess of concentrated sulphuric acid at 443 K gives ethene.

CH₃CH₂OH, ethanol

CH₃CH₂OH
$$\xrightarrow{\text{conc.H}_2SO_4}$$
 CH₂=CH₂ + H₂O

Ethanol Ethene

Answer.

4. What is meant by a sturated hydrocarbon?

Answer. Those hydrocarbons in w hich valency of carbon is satisfied by single bonds only are called sturated hydrocarbons.

5. Name the compound formed when ethanol is warmed with ethanoic acid in the presence of a few drops of cone.H₂S₀₄

Answer.

Ethyl ethanoate is formed.

$$CH_3COOH + C_2H_5OH \xrightarrow{conc.H_2SO_4} CH_3COOC_2H_5 + H_2O$$

Ethanoic acid Ethanol Ethyl ethanoate Water

6. Draw the structure of CH₃COOH molecule.

Answer.

7. Draw the structure of ethanol molecule.

Answer.

8. What happens when a small piece of sodium is dropped into ethanol?

Answer.

Hydrogen gas will be evolved.

$$2C_2H_5OH(l) + 2Na(s) \longrightarrow 2C_2H_5ONa(l) + H_2(g)$$

Carbon has four electrons in its valence shell. How does carbon attain stable electronic configuration.

Answers. By sharing four electrons with other atoms.

10. State two characteristic features of carbon which when put together give rise to large number of carbon compounds.

Answer. (i) Catenation (ii) Tetravalency of carbon

11. Write the structural formula of chloroethane.

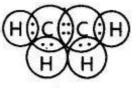
Answer

12. How many covalent bonds are there in a molecule of ethane (C₂H₆)?

Answer. There are 7 covalent bonds in a molecule of ethane.

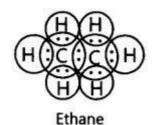
13. Write the electron dot structure of ethene molecule (C₂H₄).

Answer.



Ethene

14. Write the electron dot structure of ethane molecule (C_2H_6). Answer.



15. Draw the structure of butanone molecule, CH₃COC₂H₅.

Answer.

16. Draw the structure of the hexanal molecule, $C_5H_{11}CHO$.

Answer.

17. Name the following compound:

Answer. 1-Hexyne is IUPAC name of the compound.